

BASEL

National Center of Competence in Research "Nanoscale Science"

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Molecular and carbon-based electronic systems

Single molecule deposition and properties on surfaces



Functional Devices



Overview

- Introduction into SPM techniques
 - interaction forces
 - detection mechanism & setup
- Properties of single C₆₀ molecules
 - orientation of single molecules
 - mechanical properties
- Manipulation of porphyrin molecules
 - structural analysis
 - 3D force spectroscopy
 - controlled molecular manipulation
- Formation of a molecular wire
 - on surface reaction
 - determination of pulling forces
- Electronic Information at submolecular scale
 - Donor and Acceptor molecules
 - Optoelectronic excitation of CuPc

MCES - FS17











Th. Glatzel, Uni Basel (2017)

Donor and acceptor molecules

4-(4-(2,3,4,5,6-pentafluorophenylethynyl)-2,3,4,5tetrafluorophehylethnyl)phenylethynylbenzene (FFPB)



Fluorine: Strong electron affinity > acceptor Non-Substituted carbon ring > donor

AADD molecule

Dipole: 4.27 Debye (i.e. 2.3 times larger than water)



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Au(110) 2×1

D. Matsuo et al., Chem. Lett. 39, 1300 (2010).

STM image – self assemble structure



Deposited on cold substrate (~120 K) Measurement at 4.8 K





¹⁰ pA, -2.0 V

Attempt to get a real conformation via DFT

DFT result show no difference



Au tip Towards seeing the chemical structure

Constant height mode



Current map



l (pA)

Frequency shift map



∆f (Hz)



Molecule tip Seeing real structure in self-assembly



NC-AFM image



DFT calculation



-Excellent agreement-

Quasi-static bias spectroscopy Charge distribution

Procedure

- Track to the marker site
- Turn off Z feedback
- Move to the measurement point
- Set tip closer to the sample by 50 pm
- Bias dependent measurement
- Move back to the marker site
- Turn on the Z feedback (-500 mV, 50 pA)



Local contact potential difference



Bias sweep : ±500 mV Points : 256

$$\Delta LPCD = 10.4 \text{ mV}$$



2D map extracted via $\Delta f(x,y,V)$



STM topography

28× 59 grid points (1652 points), 16hours33mintues, restricted by the refilling cycle of liquid He C-H and C-F induce a net dipole moment of 4.27D ($H_2O \sim 2D$) along the molecular axis

CuPc molecules on Cu surfaces related publications



Th. Glatzel, Uni Basel (2017)

H. Karacuban et al., Surf. Sci. 603, L39, (2009). J. Schaffert et al., Nature Mat. 12, 223, (2013). J. Schaffert et al., PRB 88, 075410, (2013).

CuPc deposition on Cu(111)

Cu-phthalocyanine (CuPc)





U=-200mV, I=30pA

CuPc deposition on Cu(111)



on Ag(111) on Cu(111)

structurally similar, higher interaction on Cu(111)

1 nm

- physisorption (vdW interactions)
- chemisorption (chem. bonds)
- adsorption geometry defined by mechanical & electronic properties

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low

1 nm

CuPc on Cu(111) Local Adsorption Geometry



U=-200mV, I_t=30pA

DFT with VdW correction and in PBE form



- strong interaction
- 6.58 eV/molecule
- symmetry reduction

CuPc on Cu(111) Local Adsorption Geometry



U=-200mV, I_t=30pA

Th. Glatzel, Uni Basel (2017)

CuPc switching



U=30 mV

U=-30 mV

3D bias spectroscopy current maps



- contrast transition in simultaneously recorded I, (x,y,U)
- different for forward and backward directions

3D bias spectroscopy current curves



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Switching Adsorption Configuration



- telegraph noise corresponds to frustrated rotations
- bistable regime: controllable switching of adsorption configuration
- can be induced upon scanning with different bias voltages

Substrate Molecule Interactions CuPc on NaCl(2ML) / Cu(111)



U=-1.7V, I_t=4pA

DFT with VdW correction and in PBE form



- weak interaction
- 2 eV/molecule
- symmetry preservation

Substrate Molecule Interactions CuPc on NaCl(2ML) / Cu(111)



No charging observed



CuPC on Cu(111) and NaCl Local Contact Potential Difference



LCPD of CuPc **Comparison with DFT calculations**

CuPc on NaCl(2ML) / Cu(111) x meas -0.01 x meas — rit 194 ____ fit -0.196 experiment x meas 196 — fit -198 20 ∑-0.198 50 −0.2 ULOPO [V] -0.03 -20040 202 -60 204 -0.0580 -0.8 -0.4 0 0.4 0.8 -0.6 -0.3 0 0.3 0.6 0.9 Lateral distance from Cu Inmi Lateral distance from Cu [nm] -0.202 -0.07 -0.204 -0.6-0.4-0.2 0 0.2 0.4 0.6 d [nm] -0.9 -0.6 -0.3 0 d [nm] 0.3 0.6 0.9 -0.25F hollow ^o bridge¤ 0.3 0.2 ∑-0.3 ⊐ ∑0.1 ⊐ 0 DFI 0 -0.1 -0.35 -0.2 -0.6 -0.4 -0.2 0 d [nm] 2 0 0.2 0.4 0.6 d [nm] 0.2 0.4 0.6 -0.6 -0.4 -0.2

CuPc on Cu(111)

A. Sadeghi et al., Phys. Rev. B., 86, 075407 (2012) A. Sadeghi, phd-thesis, University of Basel (2013)

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1D bias spectroscopy CuPC-tip on C₆₀ on Cu(111): under illumination



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Overview 2

Kelvin Probe Force Microscopy

- Measurement principle
- Experimental setup

• Cyano-Porphyrin Wires

- Growth along step edges of KBr
- Multiwire assemblies on NaCl and KBr
- Contacting and cutting molecular wires

Truxenes

- Self assemblies on KBr crystals
- Molecular structures on patterned surfaces
- Reconstructing surfaces
- Single molecule at room temperature







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noncontact Atomic Force Microscopy nc-AFM / KPFM principle



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Th. Glatzel, Uni Basel (2017)

Kelvin Probe Force Microscopy

Principle - biomodal detection (AM-KPFM)



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Experimental Setup nc-AFM and KPFM



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Experimental Results: nc-AFM

inhomogeneous sample: HOPG + $\frac{1}{2}$ monolayer C₆₀



 \rightarrow contrast inversion: HOPG \leftrightarrow C₆₀

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S. Sadewasser et al., PRL, 2003, 91, 266101

Experimental Results: KPFM

inhomogeneous sample: HOPG + ¹/₂ monolayer C₆₀



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S. Sadewasser et al., PRL, 2003, 91, 266101

Capacitive Cross Talk bias-spectroscopy on KBr



V_{AC}=1V, f₂=960.831kHz

Atomic Scale Contrast in AM-KPFM

Truxenes on KBr



Si(111)



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Cyano-porphyrins on 1MI KBr



Au & porphyrins on Cu(111)/KBr



A. Sadeghi et al., Phys. Rev. B 86, 075407, (2012).
G. Elias et al., Beilstein J. of Nanotech. 2, 252-260, (2011).
S. Kawai et al., Nanotechnology 21, 245704, (2010).
L. Nony et al., Nanotechnology 20, 264014, (2009).
Th. Glatzel et al., Nanotechnology 20, 264016, (2009).
G. Enevoldsen et al., Phys. Rev. Lett. 100, 236104, (2008).
F Bocquet et al., Phys. Rev. B 78, 035410, (2008).

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Molecules on Insulators:

- No STM possible nc-AFM mandatory
- Low diffusion barrier but high intermolecular interaction
- Low temperatures easier to "fix" molecules but not easy to find applications

Asymmetric Cyano-Porphyrins Structure and Wire Formation



- able to $\pi \pi$ stack
- negative charge at the nitrogen atom induces a dipole (p ~ 4.37 D)
- two 3,5-di(tert-butyl)phenyl- groups act as spacers
- formation of mono-molecular wires
- structure growth across terraces

Wire Formation at step edges of KBr(001)



F. Cheng et al. Chem. Eur. J. 12, 6062-6070, (2006). Th. Glatzel et al., Beilstein J. Nanotechnol. 2, 34-39, (2011).

Wire Formation Structural model



- tilt angle is determined by the side groups, the π - π stacking and the step height
- Steps higher than 3 ML prevent a π - π stacking



Th. Glatzel et al., Beilstein J. Nanotechnol. 2, 34-39, (2011).

Molecular Assemblies Multiwires on KBr



- Multiwire growth across terraces
- The <110> directions are preferred
- Different heights are visible

Molecular Assemblies High resolution imaging



Incommensurate growth in <110>

Distance between K⁺ ions:

<110>: 4.65 Å <100>: 6.60 Å

Molecular Assemblies Structural model



- Inter-molecular equilibrium separation ~ 5.7 Å
- Directed growth by the substrate
- Distance between Na⁺ ions: <110>: 3.99 Å
 <100>: 5.65 Å
- Distance between K^+ ions: <110>: 4.67 Å

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Th. Glatzel et al., Beilstein J. Nanotechnol. 2, 34-39, (2011).

<100>: 6.60 Å

Contacting Molecular Assemblies Au-Molecules-Au



- Molecules arrange at steps and across terraces
- The growth is started/stopped at gold clusters.

Interface of Molecules and Au Topography and Surface Potential



- 250 mV between the KBr surface and the Au nanoclusters
- 220 mV between Au nanocluster and the molecular wire

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Th. Glatzel et al., APL 94, 063303 (2009)

Self-Healing of Molecular Wires

Topography



Parameter: 90x90nm², A = 5nm, γ = -0.5fN \sqrt{m} , V_{bias} = 0.43V

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S. Kawai, Th. Glatzel et al., APL 95, 103109 (2009)

Contacting Molecular Assemblies Nanostencil (IBM Rüschlikon)







L. Gross, Th. Glatzel et al., J. Vac. Sci. Technol. B, 28, C4D34-C4D39, (2010).

Contacting Molecular Assemblies Nanostencil (IBM Rüschlikon)



300x300nm²

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Truxenes on Patterned Surface

filled and unfilled pits measured at RT

- cooperation with A. Echavarren, Tarragona
- molecules has three CN groups
- better sticking to ionic surfaces expected





and the result of post annealing at 155 C for 15 mins





Truxenes on patterned surface Filled and unfilled pits



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Th. Glatzel, Uni Basel (2017)

Reconstruction of Ionic Surfaces Truxene molecules



T. Trevethan, Th. Glatzel et al., Small 7, 1264, (2011)

Imaging a Single Molecule Measurements at RT and Quantum Chemical Calculations



binding energies: kink: 1.33 eV step: 1.01 eV MCES - FS17 surface: 0.42 eV





B. Such, Th. Glatzel et al. ACS Nano, 4, 3429-3439, (2010).

Calculations of adsorbed Truxenes DFT calculations and MD simulations



Potential Energy Change Transforming the Island/Pit Structure



Potential energy to create a pair of kinks: - 0.44eV Decoration by two or more truxene molecules: + 0.64eV

3D dynamic force spectroscopy at RT DPDI molecular network

Cooperation with L. Gade, Th. Jung and M. Stöhr



M. Stöhr; Angew. Chem. Int. Ed. 44, 7394(2005)

$$A_{2nd} = 400 \text{ pm}$$
 $A_{TR} = 50 \text{ pm}$



8 nm x 8 nm

3D dynamic force spectroscopy at RT DPDI molecular network



Th. Glatzel, Uni Basel (2017)

Conclusion

opto-electronic charge transfer processes in molecules



- locale surface potential at atomic scale surface photovoltage
- transfer to room temperature
- stabilization and **manipulation** of molecules/atoms
- quantification of the observed signals (forces and energy)
- development of new measurement methods