



Molecular and carbon-based electronic systems

Lecture 5 Molecular junctions basics

Vorlesung Uni Basel, HS2017

- single molecule electronics
- atomic contacts and conductance quantization
- contacting individual molecules : mechanically controllable break junctions
 - formation mechanisms: fluctuations, conductance plateaus
 - understanding transport through junctions: Landauer approach
 - conjugated compounds
- examples of spectroscopy in molecular junctions
 - current-voltage characteristics
 - conducting AFM
- outlook



(Volume reduced by 10¹²)

Molecular and carbon-based electronic systems



1940s Robert S. Mulliken: concept of donor-acceptor charge transfer complexes



Albert Szent-Gyorgi:

proteins "might not be insulators"

historical perspective



1960s Eley and Spivey (*Trans. Faraday Soc., 1962*) *"It seemed therefore reasonable to suppose that a DNA molecule might behave as a 1D aromatic crystal and show a p-electron conductivity down the axis."*

"...if it should prove possible to measure a single fiber..."

1970sB. Mann and H. Kuhn (J. Appl. Phys., Oct. 1971)1st reproducible transport meas. through organic layers

A. Aviram, M. A. Ratner: **molecular rectifier** (theory) *Chemical Physics Letters 29 (2), Nov.1974* from molecular conduction to **molecular electronics**



nb citations of Aviram, Ratner paper

exponential development

1985's - 2000's



The Nobel Prize in Chemistry 1987

"for their development and use of molecules with structure-specific interactions of high selectivity"





Donald J. Cram

Jean-Marie Charles J. Pedersen

Lehn



The Nobel Prize in Chemistry 1996 "for the discovery of fullerenes"





Sir Harold

W. Kroto



Robert F. Curl Jr.

Richard E. Smallev



The Nobel Prize in Chemistry 2000

"for the discovery and development of conductive polymers"



The Nobel Prize in Chemistry 1992

"for for his contributions to the theory of electron transfer reactions in chemical systems "



Rudolph A. Marcus



Alan G. MacDiarmid Hideki Shirakawa Professor at the University of Pennsylvania, Philadelphia, USA, University of Tsukuba. Japan.

Alan J. Heeger Professor Emeritus, Professor at the University of California at Santa Barbara. USA.

conducting polymers





conducting polymers



conducting polymers



nobelprize.org



TV display 40" OLED (EPSON, 2004)



press release, Nobel prize 2000

(...) In the future we will be able to produce transistors and other *electronic components consisting of individual molecules* - which will dramatically increase the speed and reduce the size of our computers. A computer corresponding to what we now carry around in our bags would suddenly fit inside a watch ...

Low-cost organic electronics on plastic Forrest, Nature (2004)









\Rightarrow at the level of a few or even a single molecule ?



reviews, e.g.: Liljeroth (2010), de Boer et al., (2008); Ratner et al; (2008); Cahen et al., (2008)

molecules: pros & cons

- molecules are small: typ. 1-100 nm
- molecules have extended pi systems

provides thermodynamically favorable electron conduit: molecules as "wires"

- molecules have **discrete energy levels** better confinement of the charges as in Si devices
- molecules can be designed/tailored

by choice of composition and geometry, the transport, optical and geometrical properties can be adjusted

• molecules are identical

chemists synthesize 1mmol of (identical) molecule at a time, not one device

• molecules can be **active** (have a function)

stereochemistry (distinct stable geometric structures – isomers), mechanical flexibility (rotation axis),
photochemistry (photochromism), electrochemistry (redox reactions)
(self-)assembly (building of structures) and molecular recognition (switching, sensing)

- reliable connection to the micro/macro-scopic world (contacts) and characterisation of a single molecule ?
- limited thermal and electrical stability
- what about the **reproducibility** of molecular devices ?
- how to fabricate/integrate many devices (upscaling) ?

nanometer and molecular-scale junctions



"playground " for

fundamental aspects

- electro-mechanical properties (e.g.: atomic, molecular switch)
- transport at μs, ns, ps, ...
- e-e and e-ph interactions
- heat flow (atomic & molecular level)
- spin dependent transport & selectivity
- exciton generation, separation
- interaction with EM field (plasmonics)
- coherence aspects only indirect evidence to date

device aspects

- control of molecule-electrode interface
- reliable 2-terminal switches (V-driven) conformational change, interference
- few molecules devices and monolayers, pores & crossbars (Sony, HP, NIST, IBM)
- carbon-based electronics
- upscaling, programmability NB: variability, tunneling, power dissipation, cost, are current issues in CMOS tech.

molecular junction



nanometer and molecular-scale junctions

- structural disorder
- interactions
- fluctuations

typ. energies

- Au-Au, Au-S bond
- surface Au-Au
- benzene-benzene

electrodes and junction geometry anchoring, self-assembly, polymerization mobility of (surface) atoms, molecular distortions, multiple local energy minima

- 0.7-1.5eV Kawai et al., APL 2008, Tao et al., JACS 2009, JL arrays
- ~ 0.3eV
- ~ 0.1eV Jorgensen et al., JACS, 1990

molecular junction



nanometer and molecular-scale junctions

- structural disorder
- interactions
- fluctuations

electrodes and junction geometry anchoring, self-assembly, polymerization mobility of (surface) atoms, molecular distortions, multiple local energy minima

- ⇒ junction formation (and breaking up): dynamic process, variability
 - time scale(s)
 - local environment effect
 - local geometry and structure effect

contacting a molecule



contacting a molecule: exp. techniques



Mantooth, Weiss., Proc. IEEE, 2003

forming atomic contacts

break junctions: forming atomic contacts





hand-made notched Au wire



Moreland & Ekin, J. Appl. Phys, (1985); Ruitenbeek et al., Phys. C (1992); Reed et al., APL (1995)

break junction: a draw bridge at the atomic scale



break junctions



elongation: $d= 6thz/L^2$ reduction factor: $r=\Delta d/\Delta z \approx (1.6 - 4) \cdot 10^{-5}$ $\Rightarrow \Delta Z = 10 \mu m \Leftrightarrow \Delta d \sim 3 \text{ Å}$

vertical speed: v_z=30µm/s ⇒ 0.5 - 1.2 nm/s electrodes separation speed

atomic-scale metallic contacts with well controlled sub-nm gap



Moreland & Ekin, J. Appl. Phys, (1985); Ruitenbeek et al., Phys. C (1992); Reed et al., APL (1995)

break junctions

closed ⇒ open: contact to tunneling



conductance quantization



 $\ell > L$ ballistic regime

Fermi wavelength (Au) GaAs: 200nm

 $\lambda_F pprox 5.2 \AA$

Energy modes splitting GaAs: ≈ 1meV $rac{\pi^2 \hbar^2}{2m \lambda_{
m F}^2} pprox 1.4 eV$

$$G = \frac{2e^2}{h} \sum_{n} T_n$$
Landauer, 1957
Landauer, Büttiker
$$I = GV = \frac{2e^2V}{h} \sum_{n} T_n$$

$$G_0 = \frac{2e^2}{h}$$

$$G_0 \simeq (12.9k\Omega)^{-1}$$

conductance of a quantum coherent structure accommodating N channels *Tn: transmission probability through channel n*

break junctions

Conductance quantization for Au contact



conductance quantization: 2DEG systems



break junctions in liquid environment



"high" conductance regime



$${
m G} \geq {
m G}_0 = 2{
m e}^2/{
m h} =$$
(12.9 k Ω) ⁻¹

conductance histogram (counts)



opening: conductance vs elongation



break junction opening (solvent:THF/mesitylene 1:4 v/v)

conductance quantization, plateaus

$$G_0 = \frac{2e^2}{h} = 77.5 \mu S$$

cf e.g. Agrait et al., Phys. Rep. 2003

- last Au G-plateau (~1G₀)
- breaking of Au bridge
- tunneling regime

$$G \propto e^{-\beta d}$$

 statistics: histogram tunneling background
 const.





tunneling regime



tunneling regime

gap open, close the junction



ø

$$I \propto \exp\left(-\frac{2d}{\hbar}\sqrt{2m\phi}\right)$$

 $d=r \cdot z$ r = reduction factor



$$\implies I \propto e^{-Bz}$$

 $\phi = U_0 - E_F$ in vacuum: metal workfunction in solvent: apparent barrier height

d

E_F

 U_0

tunneling regime: attenuation factor



purely elastic: $r=6 t h/L^2 \sim 5 \times 10^{-6}$



"hinge" model: $r\approx 4~d/L$ ~ $6~x~10^{-4}$

In vacuum

$$\phi_{vac} = 3.5 - 5 \text{ eV} \Rightarrow r \sim 5 \times 10^{-5}$$

 $B = 2r\sqrt{2m\phi}/\hbar$ $I \propto e^{-Bz}$

NB: plastic deformation of the substrate can affect r

contacting simple molecules

dynamical behavior & fluctuations: solvent



J. Brunner et al., to appear in JPCM, 2014

dynamical behavior & fluctuations: solvent + alkanedithiols



J. Brunner et al., to appear in JPCM, 2014

dynamical behavior & fluctuations: solvent + alkanedithiols



formation of a molecular junction: opening traces





only solvent

signature of a molecular junction



example: alkanedithiols



large HOMO-LUMO gap (few eV), ~ no solvent effect

molecular signature:

plateau (at G « G₀) (Reichert et al., PRL 2002, Tao et al., Science, 2003, Nano Lett. 2004, JACS 2007)



only solvent: Mesitylene with molecules:0.1 M octanedithiol in Mesitylene

signature of a molecular junction



model system: alkanedithiols



large HOMO-LUMO gap (few eV), ~no solvent effect

molecular signature: plateau (at G ≪ G₀) ⇔ peak in (log-G) histogram



- plateau in 20-60% curves
- conductance window > 6 orders of mag.

M.T. Gonzàlez et al., Nano Lett. (2006) & New J. Phys. (2008)

statistical analysis



values at 1.2, 4.5 and 25 10^{-5} G₀



interpretation of the different values

- anchoring geometry
- conformational states
- multiple bridges



T. Gonzàlez et al., New J. Phys. (2008)

dynamical behavior & fluctuations: solvent + alkanedithiols

